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<p>(54) Title: SUPPORTED METALLOCENE CATALYST SYSTEMS FOR THE POLYMERIZATION OF OLEFINS, PREPARATION AND USE THEREOF</p> <p>(57) Abstract</p> <p>The invention relates to supported metallocene catalyst systems, optionally prepolymerized with olefinic monomer, preferably ethylene or propylene, useful in the polymerization of propylene to isotactic polypropylene. The isotactic polypropylene obtained generally has a molecular weight of about 100000 g/m or greater, a melting point of about 140 °C or greater and granular morphology. The supported catalyst system generally has high catalytic activity, and upon use in a polymerization reactor results in a process with minimal to no reactor fouling, fining, or chunking.</p>		

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**SUPPORTED METALLOCENE CATALYST SYSTEMS**  
**FOR THE POLYMERIZATION OF OLEFINS,**  
**PREPARATION AND USE THEREOF**

5

**FIELD OF THE INVENTION**

The invention relates to catalyst systems and more particularly to prepolymerized supported metallocene/alumoxane catalyst systems useful for the polymerization of olefins or mixtures of olefins to polyolefins. It is especially  
10 useful for the production of granular isotactic polymers of propylene of high molecular weight and high melting point.

**BACKGROUND OF THE INVENTION**

Supported catalyst systems are well known.

15 Techniques to support a catalyst system vary widely. For example techniques employing hydrated carriers or supports are disclosed in EPA 367503, EPA 336593, and US 4912075; techniques employing varying types of inert carriers and metallocenes are disclosed in US 4659685. US 4808561 and US 4701432 describe techniques wherein the carrier is calcined and subsequently  
20 contacted with the metallocene and activator or cocatalyst component.

Methods are known in the art to obtain polymer product having uniform compact spherical particles, narrow particle size distribution and/or high bulk density with supported catalysts. EPA 302424 and EPA 354893 describe preactivating or precontacting the metallocene and activator; EPA 426646 and US  
25 4871705 describe prepolymerizing a supported catalyst system in the presence of at least one olefin; EPA 279586 describes utilizing finely divided alumoxane to yield a powdery polymer; and EPA 314797 describes utilizing a supported catalyst system and fillers in the system with affinities for polyolefins.

Prior art supporting techniques generally produce a catalyst system whose  
30 catalytic performance is inferior compared to its unsupported counterpart. The activity of the resulting supported catalyst is less than the activity of the unsupported catalyst system. The resultant polymer generally has a lower molecular weight and/or lower melting point. Any advantages gained with respect to polymer morphology employing prior art supported catalyst had to be balanced  
35 against the counterproductive effects due to fouling, fining, sheeting and chunking

which generally were observed in the polymerization reactor of gas phase, bulk or slurry polymerization processes.

It is well known that lower polymerization reaction temperatures could raise melting point and molecular weight of the polymer produced, but lowering the polymerization reaction temperature is not always practical. Supported catalyst systems which can operate at high reaction temperatures, i.e., those greater than 60°C, are commercially attractive since they generally allow for increased catalyst activity and the absence of special refrigeration conditions to remove the resultant heat of reaction liberated during polymerization. When employing supported catalysts, considerations such as catalyst activity, and polymerization reaction conditions as well as the resultant polymer molecular weight and melting point must be taken into account.

Homogeneous or nonsupported metallocene catalyst systems are known for use in the production of high molecular weight, high melting point, narrow molecular weight distribution stereoregular polymers. In particular, substituted bisindenyl zirconocenes and alumoxane activators have been found to be suitable catalyst systems for the production of isotactic, high melting point, high molecular weight, narrow molecular weight distribution polymer. Some of these catalyst systems have been found to be highly active catalyst systems. As discussed above however, once placed onto inert carriers, these catalyst systems tend to suffer loss of activity and the ability to produce polymer of high molecular weight and/or high melting point.

To summarize the background of the invention, there exist metallocene catalyst systems which are highly active and produce high molecular weight, high melting point, narrow molecular weight distribution stereoregular polymer. There exist supporting techniques to place metallocenes on inert carriers. However, when employing prior art supporting techniques with prior art highly active catalyst systems for production of stereoregular polymers, a supported catalyst system results which exhibits poor catalytic activity compared to its nonsupported counterpart. The prior art catalyst systems generally produce a polymer with decreased melting point, and molecular weight relative to that produced with the analogous nonsupported catalyst system.

A need exists to develop a supported catalyst system which produces a granular, highly isotactic polyolefin having a high melting point of about 140°C or greater, a molecular weight of about 100000 or greater and a narrow molecular

weight distribution. Molecular weight is defined herein as weight average molecular weight, and molecular weight distribution (MWD) is defined as the ratio of weight average molecular weight to number average molecular weight ( $M_w/M_n$ ). In order to be commercially attractive, the catalyst system must produce little to no reactor fouling or sheeting and maintain its stability over a sufficient period of time to allow for storage and transport of the supported catalyst if necessary.

### **SUMMARY OF THE INVENTION**

The present invention relates to group 4, 5, or 6 transition metal prepolymerized supported metallocene/ alumoxane catalyst systems, being useful for the polymerization of propylene to isotactic polypropylene or propylene copolymers. The granular isotactic polypropylene, homo- or copolymers thereof, generally have a narrow molecular weight distribution of about 2.5 or less, a melting point of about 140°C or greater, a molecular weight of about 100000 grams/mole (g/m) or greater, a bulk density of about 0.30 g/cm<sup>3</sup> or greater and an average particle size of about 500 to 1000 microns or greater. The inventive supported catalyst can be employed in gas phase, bulk, solution, or slurry polymerization processes, at reaction temperatures in excess of about 45°C.

Catalyst systems supported by the inventive process have been found to retain catalytic performance or properties similar to their unsupported counterpart. The performance characteristics include activity, the ability to produce polymer with melting point, molecular weight and morphology similar to that produced with the unsupported counterpart. Additionally, the supported catalyst systems were found to produce little to no fouling, fines, or chunking in the reactor.

### **DESCRIPTION OF PREFERRED EMBODIMENTS**

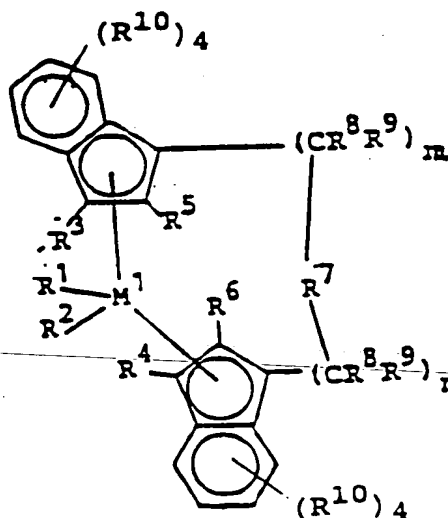
The present invention provides a supported catalyst system and method to make and use the same, comprising a metallocene, preferably a biscyclopentadienyl derivative of a group 4, 5, or 6 transition metal, an aluminum alkyl activator such as alumoxane, and a porous, preferably inorganic support. The invention more specifically provides a supported catalyst system comprising a reaction product of one or more metallocenes, as described herein, and alumoxane, a porous support, and, optionally prepolymerized olefinic monomer(s). A process for preparing these catalyst systems is provided, as well as a process for forming homo- or copolymers

of isotactic polypropylene from the catalyst system. The supported catalyst system is especially useful in the production of polymers of propylene, particularly granular isotactic polypropylene having high molecular weight, high melting point, and a bulk density in the range of about 0.30 to about 0.45 g/cm<sup>3</sup> or greater. The resulting isotactic polypropylene is free flowing, and easy to store and transport.

As used herein the terms "carrier" and "support" are interchangeable and are intended to refer to any structure which can serve as a foundation for maintaining a catalyst component or catalyst system. Metallocene is defined as a derivative of a pi-bonding moiety such as a cyclopentadienyl (Cp) moiety or derivative thereof, or an ancillary ligand which will coordinate with a transition metal. The transition metal employed is a group 4, 5, or 6 metal, preferably group 4, and most preferably is zirconium (Zr). The Periodic Table of Elements referred to herein is that as described as the New IUPAC notation in Hawley's Condensed Chemical Dictionary, Eleventh Edition, revised by Sax and Lewis, Van Nostrand Reinhold, New York, 1987.

#### A. Metallocene Component

Metallocene components are broadly disclosed in U.S. Patent 5,017,714 and EP-A-129 368. The metallocene components employed in the present invention comprise group 4, 5, or 6 transition metal, biscyclopentadienyl derivatives, preferably bis-indenyl metallocene components having the following general structure:



wherein:  $M^1$  is a metal of group 4, 5, or 6 of the Periodic Table, for example titanium, zirconium, hafnium, vanadium niobium, tantalum, chromium, molybdenum and tungsten, preferably, zirconium, hafnium and titanium, most preferably zirconium;

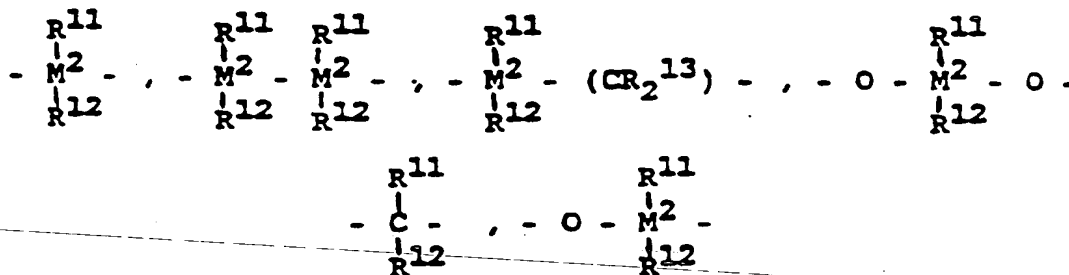
5  $R^1$  and  $R^2$  are identical or different, are one of a hydrogen atom, a  $C_1$ - $C_{10}$  alkyl group, preferably a  $C_1$ - $C_3$  alkyl group, a  $C_1$ - $C_{10}$  alkoxy group, preferably a  $C_1$ - $C_3$  alkoxy group, a  $C_6$ - $C_{10}$  aryl group, preferably a  $C_6$ - $C_8$  aryl group, a  $C_6$ - $C_{10}$  aryloxy group, preferably a  $C_6$ - $C_8$  aryloxy group, a  $C_2$ - $C_{10}$  alkenyl group, preferably a  $C_2$ - $C_4$  alkenyl group, a  $C_7$ - $C_{40}$  arylalkyl group, preferably a  $C_7$ - $C_{10}$  arylalkyl group, a  $C_7$ - $C_{40}$  alkylaryl group, preferably a  $C_7$ - $C_{12}$  alkylaryl group, a  $C_8$ - $C_{40}$  arylalkenyl group, preferably a  $C_8$ - $C_{12}$  arylalkenyl group, or a halogen atom, preferably chlorine;

$R^3$  and  $R^4$  are hydrogen atoms;

15  $R^5$  and  $R^6$  are identical or different, preferably identical, are one of a halogen atom, preferably a fluorine, chlorine or bromine atom, a  $C_1$ - $C_{10}$  alkyl group, preferably a  $C_1$ - $C_4$  alkyl group, which may be halogenated, a  $C_6$ - $C_{10}$  aryl group, which may be halogenated, preferably a  $C_6$ - $C_8$  aryl group, a  $C_2$ - $C_{10}$  alkenyl group, preferably a  $C_2$ - $C_4$  alkenyl group, a  $C_7$ - $C_{40}$  arylalkyl group, preferably a  $C_7$ - $C_{10}$  arylalkyl group, a  $C_7$ - $C_{40}$  alkylaryl group, preferably a  $C_7$ - $C_{12}$  alkylaryl group, a  $C_8$ - $C_{40}$  arylalkenyl group, preferably a  $C_8$ - $C_{12}$  arylalkenyl group, a -  
20  $NR_2^{15}$ ,  $-SR^{15}$ ,  $-OR^{15}$ ,  $-OSiR_3^{15}$  or  $-PR_2^{15}$  radical, wherein  $R^{15}$  is one of a halogen atom, preferably a chlorine atom, a  $C_1$ - $C_{10}$  alkyl group, preferably a  $C_1$ - $C_3$  alkyl group, or a  $C_6$ - $C_{10}$  aryl group, preferably a  $C_6$ - $C_9$  aryl group;

$R^7$  is

25



30

35  $=BR^{11}$ ,  $=AIR^{11}$ ,  $-Ge-$ ,  $-Sn-$ ,  $-O-$ ,  $-S-$ ,  $=SO$ ,  $=SO_2$ ,  $=NR^{11}$ ,  $=CO$ ,  $PR^{11}$ , or  $=P(O)R^{11}$ ;

wherein:

$R^{11}$ ,  $R^{12}$  and  $R^{13}$  are identical or different and are a hydrogen atom, a halogen atom, a  $C_1$ - $C_{20}$  alkyl group, preferably a  $C_1$ - $C_{10}$  alkyl group, a  $C_1$ - $C_{20}$  fluoroalkyl group, preferably a  $C_1$ - $C_{10}$  fluoroalkyl group, a  $C_6$ - $C_{30}$  aryl group, preferably a  $C_6$ - $C_{20}$  aryl group, a  $C_6$ - $C_{30}$  fluoroaryl group, preferably a  $C_6$ - $C_{20}$  fluoroaryl group, a  $C_1$ - $C_{20}$  alkoxy group, preferably a  $C_1$ - $C_{10}$  alkoxy group, a  $C_2$ - $C_{20}$  alkenyl group, preferably a  $C_2$ - $C_{10}$  alkenyl group, a  $C_7$ - $C_{40}$  arylalkyl group, preferably a  $C_7$ - $C_{20}$  arylalkyl group, a  $C_8$ - $C_{40}$  arylalkenyl group, preferably a  $C_8$ - $C_{22}$  arylalkenyl group, a  $C_7$ - $C_{40}$  alkylaryl group, preferably a  $C_7$ - $C_{20}$  alkylaryl group or  $R^{11}$  and  $R^{12}$ , or  $R^{11}$  and  $R^{13}$ , together with the atoms binding them, can form ring systems;

$M^2$  is silicon, germanium or tin, preferably silicon or germanium, most preferably silicon;

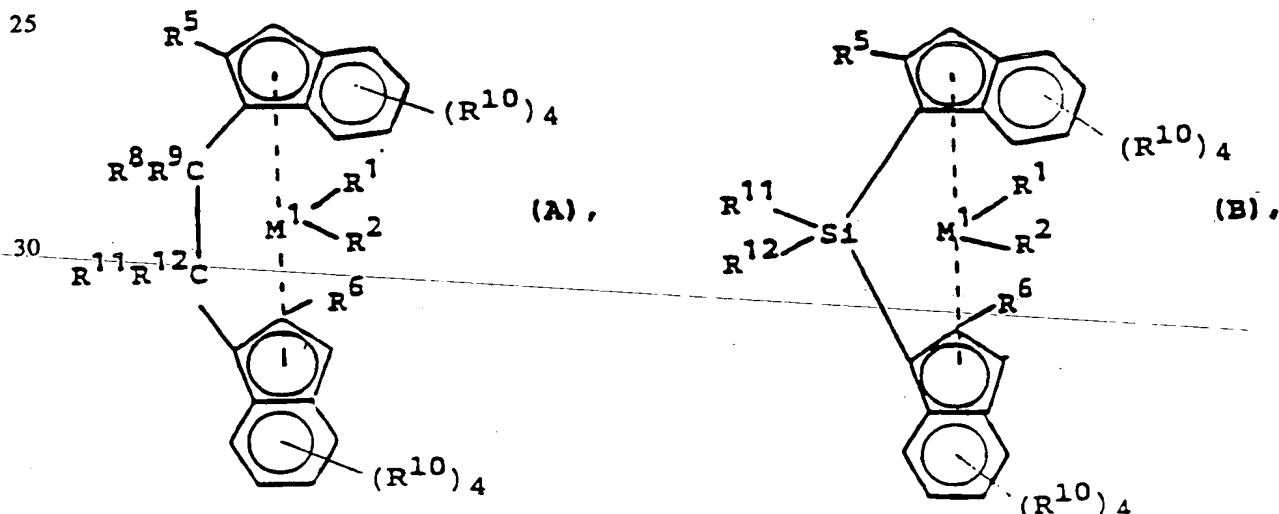
$R^8$  and  $R^9$  are identical or different and have the meanings stated for  $R^{11}$ ;

$m$  and  $n$  are identical or different and are zero, 1 or 2, preferably zero or 1,  $m$  plus  $n$  being zero, 1 or 2, preferably zero or 1; and

the radicals  $R^{10}$  are identical or different and have the meanings stated for  $R^{11}$ ,  $R^{12}$  and  $R^{13}$ . Two adjacent  $R^{10}$  radicals can be joined together to form a ring system, preferably a ring system containing from about 4-6 carbon atoms.

Alkyl refers to straight or branched chain substituents. Halogen (halogenated) is fluorine, chlorine, bromine or iodine atoms, preferably fluorine or chlorine.

Particularly preferred metallocenes are compounds of formulae:





wherein:

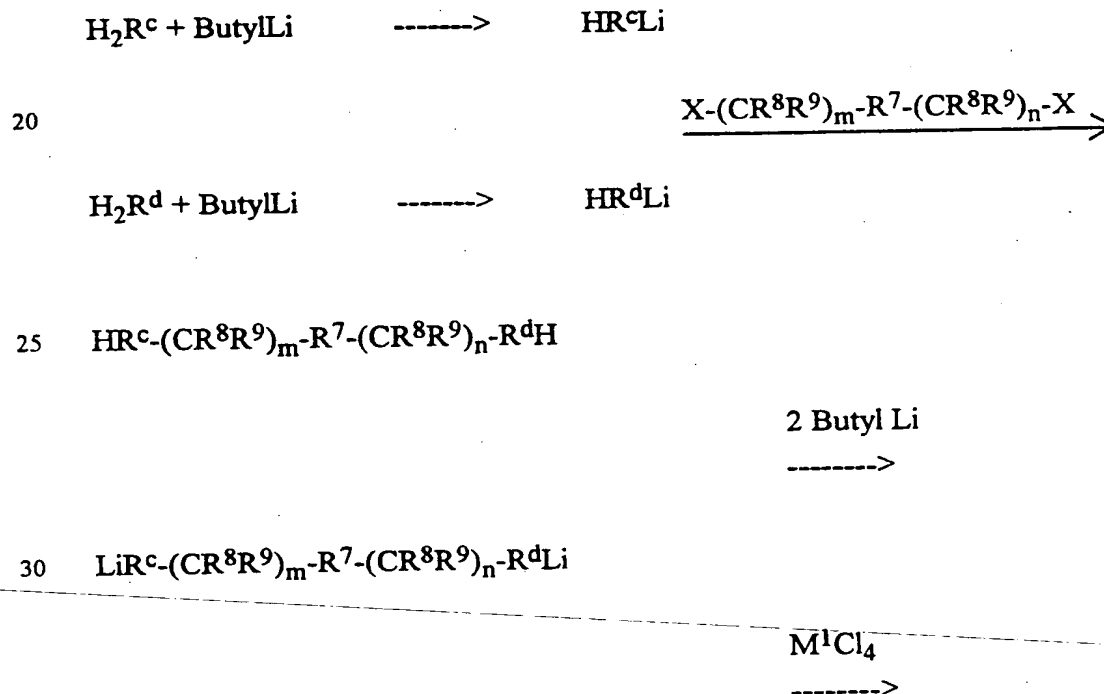
$M^1$  is Zr or Hf,  $R^1$  and  $R^2$  are methyl or chlorine, and  $R^5, R^6, R^8, R^9, R^{10}, R^{11}$  and  $R^{12}$  have the above-mentioned meanings.

The chiral metallocenes are used as a racemate for the preparation of highly isotactic poly-1-olefins.

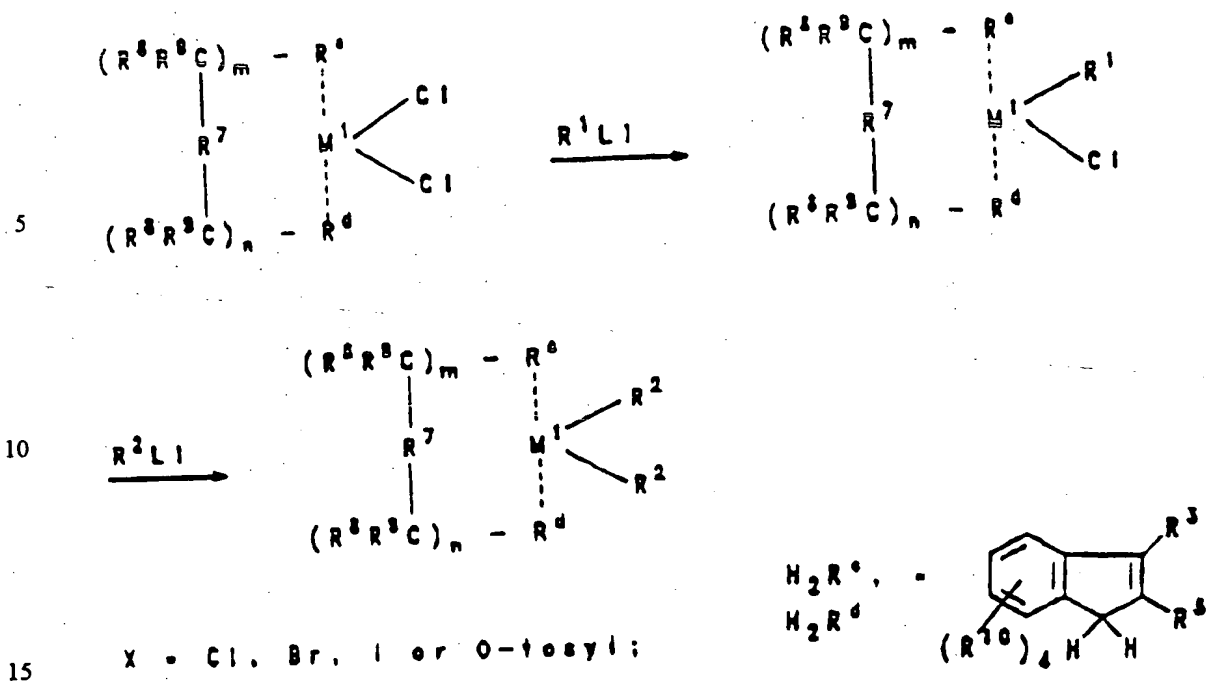
It is also possible to use the pure R or S form. An optically active polymer can be prepared with these pure stereoisomeric forms. It is preferred that the meso form of the metallocenes be removed to ensure the center (i.e., the metal atom) provides stereoregular polymerization.

Separation of the stereoisomers can be accomplished by known literature techniques. For special products it is also possible to use rac/meso mixtures.

Generally, the metallocenes are prepared by a multi-step process involving repeated deprotonations/metallations of the aromatic ligands and introduction of the bridge and the central atom by their halogen derivatives. The following reaction scheme illustrates this generic approach:



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The reader is referred to the illustrative Examples herein, the Journal of Organometallic Chem., 288, (1958), 63-67, and EP-A- 320762, for preparation of the metallocenes described. JOCS and the EPA '762 are herein incorporated by reference in their entirety.

Illustrative but non-limiting examples of metallocenes include:

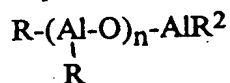
- Dimethylsilandiylbis(2-methyl-4-phenyl-1-indenyl)ZrCl<sub>2</sub>
- Dimethylsilandiylbis(2-methyl-4,5-benzoindenyl)ZrCl<sub>2</sub>;
- 25 Dimethylsilandiylbis(2-methyl-4,6-diisopropylindenyl)ZrCl<sub>2</sub>;
- Dimethylsilandiylbis(2-ethyl-4-phenyl-1-indenyl)ZrCl<sub>2</sub>;
- Dimethylsilandiylbis(4-naphthyl-1-indenyl)ZrCl<sub>2</sub>,
- Phenyl(Methyl)silandiylbis(2-methyl-4-phenyl-1-indenyl)ZrCl<sub>2</sub>,
- Dimethylsilandiylbis(2-methyl-4-(1-naphthyl)-1-indenyl)ZrCl<sub>2</sub>,
- 30 Dimethylsilandiylbis(2-methyl-4-(2-naphthyl)-1-indenyl)ZrCl<sub>2</sub>,
- Dimethylsilandiylbis(indenyl)ZrCl<sub>2</sub>,
- Dimethylsilandiylbis(2-methyl-4,5-diisopropyl-1-indenyl)ZrCl<sub>2</sub>,
- Dimethylsilandiylbis(2,4,6-trimethyl-1-indenyl)ZrCl<sub>2</sub>,
- Phenyl(Methyl)silandiylbis(2-methyl-4,6-diisopropyl-1-indenyl)ZrCl<sub>2</sub>,
- 35 1,2-Ethandiylbis(2-methyl-4,6-diisopropyl-1-indenyl)ZrCl<sub>2</sub>,

- 1,2-Butandiylbis(2-methyl-4,6-diisopropyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-methyl-4-ethyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-methyl-4-isopropyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-methyl-4-t-butyl-1-indenyl)ZrCl<sub>2</sub>,  
5 Phenyl(Methyl)silandiylbis(2-methyl-4-isopropyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-ethyl-4-methyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2,4-dimethyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-methyl-4-ethyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-methyl- $\alpha$ -acenaphth-1-indenyl)ZrCl<sub>2</sub>,  
10 Phenyl(Methyl)silandiylbis(2-methyl-4,5-benzo-1-indenyl)ZrCl<sub>2</sub>,  
Phenyl(Methyl)silandiylbis(2-methyl-4,5-(methylbenzo)-1-indenyl)ZrCl<sub>2</sub>,  
Phenyl(Methyl)silandiylbis(2-methyl-4,5-(tetramethylbenzo)-1-indenyl)ZrCl<sub>2</sub>,  
Phenyl(Methyl)silandiylbis(2-methyl-a-acenaphth-1-indenyl)ZrCl<sub>2</sub>,  
1,2-Ethandiylbis(2-methyl-4,5-benzo-1-indenyl)ZrCl<sub>2</sub>,  
15 1,2-Butandiylbis(2-methyl-4,5-benzo-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-methyl-4,5-benzo-1-indenyl)ZrCl<sub>2</sub>,  
1,2-Ethandiylbis(2,4,7-trimethyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-methyl-1-indenyl)ZrCl<sub>2</sub>,  
1,2-Ethandiylbis(2-methyl-1-indenyl)ZrCl<sub>2</sub>,  
20 Phenyl(Methyl)silandiylbis(2-methyl-1-indenyl)ZrCl<sub>2</sub>,  
Diphenylsilandiylbis(2-methyl-1-indenyl)ZrCl<sub>2</sub>,  
1,2-Butandiylbis(2-methyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-ethyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-methyl-5-isobutyl-1-indenyl)ZrCl<sub>2</sub>,  
25 Phenyl(Methyl)silandiylbis(2-methyl-5-isobutyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2-methyl-5-t-butyl-1-indenyl)ZrCl<sub>2</sub>,  
Dimethylsilandiylbis(2,5,6-trimethyl-1-indenyl)ZrCl<sub>2</sub>, and the like.

A similar list of illustrative bis indenyl species comprising various other metals could be made, but since the lists would be nearly identical to that already  
30 presented, such lists are not deemed essential to a complete disclosure. It is  
apparent to those skilled in the art that other cyclopentadienyl derivatives, and  
transition metals other than zirconium, would be useful in the compositions of this  
invention.

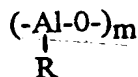
B. Activator Component

The activator or cocatalyst component of the present invention is alumoxane represented by the general formula:



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for oligomeric, linear alumoxane and



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for oligomeric, cyclic alumoxane, wherein n and m = 1-40, most preferably 3-20, and R is a C<sub>1-8</sub> alkyl group or R is an C<sub>6-18</sub> aryl group, or hydrogen, preferably a methyl group, or R can be mixtures of alkyl and aryl substituents.

15

The precise structure of alumoxane is unknown. Alumoxane or methylalumoxane can be prepared by a variety of known processes. Whatever process employed to produce the alumoxane, a varying content of unconverted original aluminum compound, in free form or as an adduct, is common to all alumoxane solutions.

C. Support Medium

Typically the support can be any organic or inorganic, inert solid, particularly, porous supports such as talc, inorganic oxides, and resinous support materials such as polyolefin or materials which may be employed as supports as described in EPA 519236.

Suitable inorganic oxide materials which are desirably employed include oxides of elements found in groups 2, 3, 4, 5, 13, 14, 15, 16 of the Periodic Table of Elements. Examples of oxides include silica, alumina, silica-alumina, and mixtures thereof. Other inorganic oxides that may be employed either alone or in combination with the silica, alumina or silica-alumina mixtures are magnesia, titanium, zirconia, and the like. Other suitable support materials can be employed such as, finely divided polyolefins, such as polyethylene.

30

If the carrier is naturally low in moisture content, or residual solvent content, dehydration or drying may not be necessary prior to use. However, if this is not the case, such as if employing a silica support, dehydration or drying is suggested. It is preferred that the carrier employed have about 1% loss on ignition (LOI) or less of water or moisture. To achieve the preferred LOI, thermal dehydration or drying treatment of the carrier may occur in vacuum or while

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purging with a dry inert gas such as nitrogen at a temperature of from about 100°C to about 1000°C, and preferably, from about 300°C to about 800°C. Pressure considerations are not critical. Generally, however, ambient conditions are employed. The duration of thermal treatment can be from about 1 to about 24 hours. However, shorter or longer times can be employed provided equilibrium is established with the surface hydroxyl groups; this is generally accomplished within 4-8 hours.

Dehydration or drying of the support can also be accomplished by subjecting the carrier to a chemical treatment in order to remove water and reduce the concentration of surface hydroxyl groups. Chemical treatment converts all water and hydroxyl groups in the oxide surface to inert species. Useful chemical agents are for example, chlorosilanes such as  $\text{SiCl}_4$ , trimethylchlorosilane, dimethylaminotrimethylsilane and the like. The chemical dehydration is accomplished by slurring the inorganic particulate material such as, silica, in an inert low boiling hydrocarbon, such as hexane. During the chemical dehydration treatment, the silica should be maintained in a moisture and oxygen free atmosphere. To the silica slurry is then added a low boiling inert hydrocarbon solution of the chemical dehydrating agent, such as, dichlorodimethylsilane. The solution is added slowly to the slurry. The temperature ranges during chemical dehydration reaction can be from about 25°C to about 120°C, however, higher and lower temperatures can be employed. Preferably, the temperature will be from about 50°C to about 70°C. The chemical dehydration procedure should be allowed to proceed until all the moisture is removed from the particulate support material as indicated by cessation of gas evolution. Normally, the chemical dehydration reaction will be allowed to proceed from about 30 minutes to about 16 hours, preferably, 1 to 5 hours. Upon completion of the chemical dehydration, the solid particulate material is filtered under a nitrogen atmosphere and washed one or more times with a dry, oxygen free inert solvent. The wash solvents as well as the diluents employed to form the slurry and the solution of chemical dehydrating agent, can be any suitable inert hydrocarbon. Illustrative of such hydrocarbons are heptane, hexane, toluene, isopentane and the like.

Characteristics that determine the amount of support to be employed in preparing the catalyst compositions, as well as affecting the properties of polymers formed include specific particle size, surface area, pore volume, and number of hydroxyl groups. Preferred carriers include silica or those having about 10-500

- 12 -

square meters/ surface area, or a porosity of about 0.2-3.5 cm<sup>3</sup> of pore volume/grams support (cm<sup>3</sup>/g).

Organic supports, such as polystyrene, may be employed and are also suggested to contain minimal to no moisture, residual solvent from processing, or impurities which may affect the final catalyst systems' performance.

D. Placing Catalyst System on Support Medium

The general support technique employed involves contacting, in a suitable solvent, a metallocene component as described above with alumoxane or methylalumoxane to form a soluble reaction product. The soluble reaction product is then contacted with the dehydrated support or carrier, the solvent removed and the resulting supported catalyst system dried to ensure essentially all or most residual solvent is removed from the pores of the carrier. A free flowing catalyst system is obtained.

A method to produce a free flowing, prepolymerized supported catalyst system thus comprises the steps of:

- a) forming in a suitable solvent, a metallocene/alumoxane mixture wherein the metallocene is as described above;
- b) contacting the mixture of (a) with a porous, generally inorganic dehydrated carrier;
- c) removing essentially all the solvent from the resulting slurry;
- d) obtaining a supported catalyst system; and
- e) optionally prepolymerizing said supported catalyst system with one or more olefinic monomer(s), thus forming a prepolymerized supported catalyst system for the production of polymers of propylene or copolymers thereof, having a molecular weight of about 100000 or greater, preferably 150000 or greater, a melting point of about 140°C or greater, preferably about 145°C or greater, and a bulk density of about 0.30 g/cm<sup>3</sup> or greater. The resulting granular polymer also has an average particle size of about 500 to about 1000 microns or greater.

At all times, the individual ingredients as well as the recovered supported catalyst component are protected from oxygen and moisture. Therefore, the reactions must be performed in an oxygen and moisture free atmosphere and the recovered supported catalyst is maintained and stored in an atmosphere which is moisture and oxygen free. Preferably, the reactions are performed in the presence of an inert dry gas such as, for example nitrogen.

Preferred solvents for the contacting of the metallocene and activator include mineral oils and the various hydrocarbons which are liquid at reaction temperatures and in which the individual ingredients are preferably soluble. Solubility of the individual components however is not necessary provided that the reaction product of the metallocene and activator is soluble in the solvent chosen. Illustrative examples of useful or suitable solvents include alkanes such as pentane, isopentane, hexane, heptane, octane, and nonane; cycloalkanes such as cyclopentane and cyclohexane; and aromatics such as benzene, toluene, ethylbenzene and diethylbenzene.

The amount of alumoxane and metallocene employed in preparation of the supported catalyst system can vary over a wide range. However, an aluminum in the activator to transition metal in the metallocene molar ratio of from about 12:1 to about 1000:1; more preferably, a ratio of about 50:1 to 500:1 is employed.

Typically 30 wt% MAO in toluene was employed in these studies, however 10 wt.% in toluene is acceptable.

In an embodiment of the present invention a toluene solution containing the metallocene is prepared to which is added a toluene solution of MAO (30 wt% in toluene). The amount of solvent should be sufficient to dissolve the reaction product, provide adequate heat transfer away from the catalyst components during the reaction and to permit good mixing. Alternatively, the same inert solvent or a different solvent may be employed to dissolve the metallocene and MAO. The metallocene solution is prepared and added to a prepared MAO solution and reacted at room temperature. The reaction between the MAO and the metallocene is rapid and it is desired that they are contacted for about one minute to one hour, preferably about ten minutes. The reaction of these two components is evidenced by its exothermic nature and a color change. However, the exotherm or color changes are not required and other analytical techniques may be employed to monitor completion of the reaction.

The soluble reaction product of the metallocene and activator are subsequently contacted with an inert carrier, typically silica, in dry form or as a slurry which slurry was prepared in the same or a different inert solvent. Preferably the silica is added dry. The ingredients can be added to the reaction vessel rapidly or slowly. Upon contact, a slurry is formed comprising the metallocene, alumoxane, and support which is preferably maintained at room temperature for about one minute to one hour, most preferably about five minutes. The

temperature maintained during contact of the metallocene/MAO solution and carrier can vary widely, such as, for example, from 0°C to 100°C. However, temperatures of greater or less than 0-100°C can also be employed. The slurry or mixture may be stirred, with or without heat and essentially all or most of the solvent removed from the supported catalyst. It is preferred to remove both the solvent which is visually observed and essentially all that in the pores of the carrier. Removal of the solvent can occur via conventional manners such as evaporation by vacuum techniques, under reduced pressure, or flashing the mixture. Heating can be maintained until the free solvent has been removed, typically about one to three hours, at a temperature from about 30°C to about 60°C. Free liquid solvent is that which is visibly present in the reaction mixture. Residual solvent is that solvent which is trapped in the pores of the support.

An alternative embodiment to removal of essentially all of the solvent is to dry the supported catalyst to a "mud" stage wherein substantially all the free liquid has been removed and subsequently wash the catalyst product with a low boiling hydrocarbon such as pentane or hexane. The supported catalyst system may then be further processed or employed in a polymerization reactor.

Although the supported catalyst may be employed as is, it is preferred to prepolymerize with one or more olefinic monomer, each having between 2 and 20 carbon atoms, preferably between 2-8 carbon atoms.

#### E. Prepolymerization of Supported Catalyst

Prepolymerization of the supported catalyst reduces reactor fouling tendencies of the catalyst and has been found to enhance particle morphology control of the final polymer formed. Although one may use olefinic monomer between about 2 and 20 carbon atoms, it is preferred that the supported catalyst be prepolymerized with ethylene or propylene monomer, most preferably ethylene. Alternatively, the supported catalyst may be prepolymerized with at least two different monomers having from 2-8 carbon atoms.

In an embodiment to prepolymerize the catalyst, the supported catalyst is slurried in isopentane or another hydrocarbon in which the metallocene or alumoxane is not soluble. Ethylene is added at a slow rate to the slurried mixture. The rate of addition is dependent upon the size of the reactor employed. Generally, when using laboratory scale reactors, the rate of addition was approximately 0.03-0.06 SCF/minute with stirring (SCF=standard cubic feet).



After a suitable period of time to allow polymerization of the ethylene monomer onto the supported catalyst, the prepolymerized catalyst is filtered through a mesh screen, decanted from the slurry mixture, washed with isopentane, dried under vacuum and the prepolymerized catalyst system is isolated. The prepolymerization typically occurs at a temperature of  $-15^{\circ}\text{C}$  to  $60^{\circ}\text{C}$ , preferably less than  $25^{\circ}\text{C}$  for about 10 minutes to about 10 hours. The amount of prepolymer can be varied and generally comprises from about 10% to about 1000%, typically from about 100% to about 300% of the supported non-prepolymerized catalyst weight.

Generally the prepolymerized catalyst resulted in a catalyst particle size of about 50 to about 100 microns. The prepolymerized catalyst produced a resultant polymer with granular morphology, generally having a particle size of about 500-1000 microns, and a bulk density in the range of about 0.30 to about 0.50  $\text{g/cm}^3$  or greater.

Generally, the more active a catalyst system, the more difficult it is to produce a supported catalyst system which retains its performance capabilities relative to its unsupported counterpart. The activity and polymerization performance of these supported bridged bis-indenyl derivative metallocenes was unexpected in view of their high activity in the unsupported format.

#### 20 F. Polymerization

All solvents were purchased from commercial sources, nitrogen purged and dried over activated molecular sieves. Aluminum alkyl reagents (MAO) were purchased as 10-30 wt% solutions in toluene from Witco Corporation, PO Box 1227, Dublin, Ohio, 43017. The silica employed was Davison 948, purchased from W. R. Grace, Davison Chemical Division, 5603 Chemical Rd., Baltimore, Md. 21226, USA, dehydrated to  $800^{\circ}\text{C}$  in a flow of nitrogen for approximately eight hours.

The prepolymerized supported catalyst product formed can be employed as a sole catalyst component for the polymerization of olefins containing from about 2 to about 20 carbon atoms, or, in the preferred alternative, it can be employed with an aluminum alkyl or alumoxane additive (or scavenger). It is preferred to employ an additive during polymerization since enhanced catalyst activity and reduced reactor fouling is generally observed. The amount of additive in the reactor is related to feed monomer purity.

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The general polymerization procedure is as follows. TEAL, used as an additive, was added to a clean, dry two liter autoclave which had been flushed with nitrogen. The reactor was closed and filled with 1000 mls of liquid propylene. The reactor was heated to the desired temperature, generally from about 65°C and the supported, prepolymerized catalyst slurried in about 2 mls of hexane, was washed in via an addition tube with approximately 250 mls propylene. After the desired reaction time, generally about 60 minutes, the reactor was cooled and the excess propylene vented. Inspection of the reactor revealed it to be clean and free of fouling, fines, or polymer chunks. The granular, free flowing polymer was removed and dried.

#### G. Isotactic Polypropylene Product

Using the prepolymerized supported catalyst system described herein, granular, free flowing isotactic polypropylene was produced comprising (a) a molecular weight of about 100000 grams/mole or greater, generally greater than 150000, (b) a melting point of from about 140°C or greater, generally greater than 145°C, (c) a bulk density of about 0.30 g/cm<sup>3</sup> or greater, (d) a narrow molecular weight distribution of about 2.5 or less and (e) a particle size of about 500-1000 microns. Homo- or co-polymers of propylene, which include isotactic polypropylene or mixtures of isotactic polypropylene and one or more other olefins, may be produced in gas phase, bulk, solution, or slurry polymerization processes, employing the supported catalyst system described, at a reaction temperature of 45°C or greater, preferably from about 60°C or greater. The resulting homo- or co-polymer of propylene can be used for preparing shaped articles, films and fibers.

The supported catalyst system in accordance with this invention provides the means to obtain highly attractive polymer product. The resultant polymers were unexpected in view of the problems generally associated with supporting highly active metallocenes.

A feature of the supported catalyst system, which is especially attractive to manufactures, is its ability to minimize or eliminate chunking, sheeting, fines generation or reactor fouling during polymerization. Additionally a granular polymer product results from its use. These features are especially attractive to manufacturers since it minimizes or eliminates the need to shut down the reaction

system for cleaning or frequent maintenance, or (2) further process the resultant polymer.

Table 1 illustrates catalyst preparation and polymerization performance of the catalyst for the production of isotactic polypropylene. The catalyst systems illustrated were prepolymerized with ethylene monomer at a temperature of from about 25-60°C. The resulting polymer had a molecular weight greater than 100000 with a melting point about or greater than 146°C. Average bulk density of the polymer obtained was about 0.45 g/cm<sup>3</sup>, with particle size greater than 500 microns.

Polymer characterizations were conducted by ASTM techniques such as DSC, GPC and elemental analysis. Melting points were determined on commercial DSC instruments and are reported as second melting point.

The advantages of the instant invention are extremely valuable from a practical and commercial standpoint. It has been demonstrated that one can support a metallocene such that the catalyst system retains much of the performance of its unsupported counterpart, and results in little to no reactor fouling during polymerization.

While this invention may be embodied in many different forms, there is described herein, in detail specific embodiments of the invention. The present disclosure, including the examples, is an exemplification of the principles of the invention and is not intended to limit the invention to the particular embodiments illustrated.

### EXAMPLES

The illustrative examples herein demonstrate the use of supported metallocene catalyst, for preparation of isotactic polypropylene.

#### A. Synthesis of rac-dimethylsilandiylbis(2-methyl-4,5-benzoindenyl)-zirconium dichloride

##### Diethyl methyl (2-naphthylmethyl) malonate (1)

5.15 g (224 mmol) of sodium were dissolved in 150 ml of absolute ethanol, while heating, and 37.3 ml (217 mmol) of diethyl methylmalonate were added at room temperature. A solution of 50 g (217 mmol) of 2-bromomethylnaphthalene (96% pure) in 270 ml of ethanol was slowly added dropwise at 0°C, and the mixture was heated under reflux for a further 4 to 5 hours. It was poured onto ice-water and extracted with ethyl acetate. The combined organic phases were dried

with sodium sulfate and evaporated. After drying under an oil pump vacuum, the oily residue was stirred with hexane at 0°C, whereupon 55 g (81%) of the compound 1 crystallized.

#### Synthesis of 2-Methyl-3-naphthylpropionic acid (2)

5 A solution of 23.7 g (422 mmol) of potassium hydroxide in 50 ml of water was added to 33.2 g (105 mmol) of the compound 1 in 70 ml of ethanol, and the mixture was heated under reflux for 4 hours. After the solvent had been stripped off, the solid residue was taken up in ethyl acetate, water was added and the pH was brought to 1 with hydrochloric acid. The aqueous phase was extracted several  
10 times with ethyl acetate. After drying over magnesium sulfate, the combined organic phases were evaporated completely. The residue was stirred with hexane for crystallization. For decarboxylation, the beige-colored solid was heated at 175°C until the evolution of gas had ended. 21 g (94%) of the product 2 were obtained as a beige-colored solid.

#### Synthesis of 2-Methyl-6,7-benzoindan-1-one (3)

22 ml of thionyl chloride were added to 21 g (98 mmol) of the compound 2, with exclusion of moisture, and the mixture was heated under reflux for 30 minutes. Excess thionyl chloride was then distilled off. The residue was briefly freed from volatile compounds under an oil pump vacuum and then dissolved in 25  
20 ml of methylene chloride, under Ar as an insert gas. The solution was slowly added dropwise to a suspension of 26 g (196 mmol) of aluminum trichloride in 60 ml of methylene chloride and the mixture was heated under reflux for a further 30 minutes. It was poured onto ice and extracted with methylene chloride. The combined organic phases were dried with sodium sulfate and evaporated. The dark  
25 oily residue was chromatographed on 600 g of silica gel 60. 8.6 g (45%) of the compound 3 were able to be eluted (yellowish solid) with a mobile phase mixture of hexane/ethyl acetate (9:3).

#### Synthesis of 2-Methyl-4,5-benzoindene (4)

2.2 g (59.5 mmol) of sodium borohydride were added in portions to a  
30 solution of 7.8 g (39.7 mmol) of the indanone, compound 3 in 400 ml of a tetrahydrofuran/methanol mixture (2:1) at room temperature, and the mixture was stirred for 14 hours. The solution was poured onto HCL-acid ice and extracted with ether. The combined organic phases were washed several times with water and dried with sodium sulfate. The orange-colored oil which remained after the  
35 solvent had been stripped off was dissolved in 240 ml of toluene, and the solution

was heated at 80°C with 570 mg (3.15 mmol) of p-toluene-sulfonic acid for 15 minutes. It was washed several times with water at room temperature, dried with sodium sulfate and evaporated. The residue was chromatographed on 300 g of silica gel 60. 4.7 g (65%) of the indene 4 were able to be eluted (colorless oil) with a mobile phase mixture of hexane/diisopropyl ether (20:1).

<sup>1</sup>H-NMR spectrum (360 MHz, CDCl<sub>3</sub>): 8.02 (1,d), 7.84 (1,m), 7.59 (1,d), 7.52 (1,d), 7.38-7.48 (2,m), 7.06 (1,m), 3.42 (2,s), 2.25 (3,d).

Synthesis of Dimethylbis(2-methyl-4, 5-benzoindenyl) silane (5)

10.2 ml (25.5 mmol) of a 2.5 M butyllithium solution in hexane were added to a solution of 4.6 g (25.5 mmol) of the compound 4 in 50 ml of tetrahydrofuran at room temperature, and the mixture was heated under reflux for 1 hour. The red solution was then added dropwise to a solution of 1.55 g (12 mmol) of dimethyldichlorosilane in 10 ml of tetrahydrofuran at room temperature, and the mixture was heated under reflux for 5 to 6 hours. The reaction solution was poured onto ice-water and extracted several times with ether. The combined organic phases were dried with sodium sulfate and evaporated, and the residue was dried under an oil pump vacuum. It was chromatographed on 300g of silica gel 60. 500 mg of unreacted starting compound 4 were initially able to be eluted with a mobile phase mixture of hexane/3% ethyl acetate. The ligand system, compound 5, then followed with the same mobile phase. After the solvent had been stripped off, this ligand system was crystallized (isomers) from hexane. The yield was 1.7 g (34%, or 44% with respect to the indene, compound 4 reacted).

Synthesis of rac - Dimethylsilanediylbis (2-methyl-4,5-benzo-indenyl) zirconium dichloride (6)

4.0 ml (10.2 mmol) of a 2.5 M butyllithium solution in hexane were added to a solution of 1.7 g (4.1 mmol) of compound 5 in 20 ml of tetrahydrofuran at room temperature under Ar as an inert gas, and the mixture was stirred at room temperature for 14 hours. The residue which remained after the solvent had been stripped off was dried using an oil pump vacuum and washed with hexane. The pale brown powder obtained was dried using an oil pump vacuum at 40 to 50°C for several hours and added to a suspension of 1.0 g (4.0 mmol) of zirconium tetrachloride in 25 ml of methylene chloride at -78°C. After the mixture had been warmed to room temperature, the solvent was stripped off and the residue was extracted with 20 ml of toluene in order to remove the meso form of the metallocene, compound 6. The residue of the toluene extract was then extracted

with 40 ml of methylene chloride. The solution was concentrated to a small volume and left to crystallize at -35°C. A total of 970 mg (42%) of the ziroconcene, compound 6 were isolated in several fractions as the pure racemate.

<sup>1</sup>H-NMR spectrum of the racemate (300 MHz, CDCL<sub>3</sub>): 7.96 (2,m), 7.78 (2,m), 7.60 (2,d), 7.48-7.56 (4,m), 7.36 (2,d), 7.27 (2,s,b-Ind-H), 2.37 (6,s,Ind-CH<sub>3</sub>), 1.36 (6,s,Si-CH<sub>3</sub>). Mass spectrum: 574 M<sup>+</sup>, correct disintegration, correct isotope pattern.

#### Supporting Catalyst Compound (6)

To an eight-liter vessel equipped with a cooling jacket and an efficient overhead stirrer was added methylalumoxane (30 wt% in toluene, 925 ml). With stirring, a suspension of compound 6 (5.0 g) in toluene (700 ml) was added under N<sub>2</sub> through a double-ended needle. After stirring for 10 minutes, dehydrated silica (Davison 948, dried at 800°C, 200 g) was added to the solution over 20 minutes. The slurry was stirred for 10 minutes and then, while a vacuum was applied from the top of the vessel, a slight flow of N<sub>2</sub> was added through the bottom. The mixture was heated to 70°C as the solvent was evaporated over a 9 hour period. The dry solid was cooled to ambient temperature overnight. Isopentane (5 liters) was added to slurry the solids and the mixture cooled to 0°C. Ethylene was added to the stirred mixture by a dip tube at a rate of 0.03-0.06 SCF/minute until a total of 491 liters of ethylene had been added. Agitation was stopped and the solids allowed to settle. The liquid was decanted from the solids, which were washed twice, each with 1.5 liters of isopentane. The wet solids were transferred to a dry-box under N<sub>2</sub> and filtered through a #14 mesh sieve. The fine particles were filtered off, washed with pentane (4 liters) and dried in vacuo. Yield: 326g.

#### Polymerization with Supported Compound 6

A sample of the supported catalyst of compound 6 (199 mg) slurried in hexane (2 ml) was flushed with propylene (250 ml) into a 2 liter autoclave, previously flushed with nitrogen, heated to 65°C and containing triethylaluminum (0.5 ml of a 1 M solution in hexane) and propylene (1000 ml). The reaction was run for 1 hour. The autoclave was cooled, vented and opened. Inspection of the interior of the reactor revealed it to be clean and free of fouling. The polymer product was dried in vacuo. 305 g of a free flowing, granular-isotactic polypropylene product were recovered. M<sub>w</sub> of 273000, bulk density of 0.43 g/cc, and an average particle size of 750 microns.

B. Synthesis of rac dimethylsilandiylbis-(2methylindenyl) - zirconium dichloride  
Synthesis of 2-Me-indene (7)

110.45 g (0.836 mol) of 2-indanone were dissolved in 500 cm<sup>3</sup> of diethyl ether, and 290 cm<sup>3</sup> of 3 N (0.87 mol) ethereal methyl Grignard solution were added dropwise at such a rate that the mixture refluxed gently. After the mixture had boiled for 2 hours under gentle reflux, it was transferred onto an ice/hydrochloric acid mixture, and a pH of 2-3 was established using ammonium chloride. The organic phase was separated off, washed with NaHCO<sub>3</sub> and sodium chloride solution and dried, yielding 98 g of crude product (2-hydroxy-2-methylindane (compound 7)), which was not purified further.

Compound 7 was then dissolved in 500 cm<sup>3</sup> of toluene, 3 g of p-toluenesulfonic acid were added, and the mixture was heated on a water separator until the elimination of water was complete and evaporated. The residue was taken up in dichloromethane and the resulting solution filtered through silica gel. The filtrate was distilled in vacuo (80°C/10 mbar). Yield: 28.49 g (0.22 mol/26%).

The synthesis of compound (7) is also described in: C.F. Koelsch, P. R. Johnson, J. Am. Chem. Soc., 65 (1943) 567-573.

Synthesis of (2-Me-indene)<sub>2</sub>SiMe<sub>2</sub> (8)

13 g (100 mmol) of 2-Me-indene (compound 7) were dissolved in 400 cm<sup>3</sup> of diethyl ether, and 62.5 cm<sup>3</sup> of 1.6N (100 mmol) n-butyllithium/n-hexane solution were added dropwise over the course of 1 hour with ice cooling. The mixture was stirred at approximately 35°C for a further 1 hour.

6.1 cm<sup>3</sup> (50 mmol) of dimethyldichlorosilane were introduced into 50 cm<sup>3</sup> of Et<sub>2</sub>O, and the lithio salt solution was added dropwise at 0°C over the course of about 5 hours, the mixture was stirred overnight at room temperature and left to stand over the weekend.

The solid which had deposited was filtered off, and the filtrate was evaporated to dryness. The product was extracted using small portions of n-hexane, and the extracts were filtered and evaporated, yielding 5.7 g (18.00 mmol) of white crystals. The mother liquor was evaporated, and the residue was then purified by column chromatography (n-hexane/H<sub>2</sub>CCl<sub>2</sub> 9:1 by volume), giving a further 2.5 g (7.9 mmol/52%) of product (as an isomer mixture).

(SiO<sub>2</sub>; n-hexane/H<sub>2</sub>CCl<sub>2</sub> 9:1 by volume) = 0.37.

R<sub>f</sub> (SiO<sub>2</sub>; n-hexane/H<sub>2</sub>CCl<sub>2</sub> 9:1 by volume) = 0.37.

The  $^1\text{H}$ -NMR spectrum exhibits the signals expected for an isomer mixture with respect to shift and integration ratio.

Synthesis of  $\text{Me}_2\text{Si}(2\text{-MeInd})_2\text{ZrCl}_2(9)$

1.68 g (5.31 mmol) of the chelate ligand dimethylsilyl-(2-methylindene) $_2$  (compound 8) were introduced into 50 cm $^3$  of THF, and 6.63 cm $^3$  of a 1.6 N (10.61 mmol) n-BuLi/n-hexane solution were added dropwise at ambient temperature over the course of about 0.5 hour. The mixture was stirred for about 2 hours at about 35°C, the solvent was stripped off in vacuo, and the residue was stirred with n-pentane, filtered off and dried. The dilithio salt obtained was added at -78°C to a suspension of 1.24 g (5.32 mmol) of  $\text{ZrCl}_4$  in 50 cm $^3$  of  $\text{CH}_2\text{Cl}_2$ , and the mixture was stirred at this temperature for 3 hours. The mixture was then warmed to room temperature overnight and evaporated. The  $^1\text{H}$ -NMR spectrum showed, in addition to the presence of some  $\text{ZrCl}_4(\text{THF})_2$ , a rac/meso mixture. After stirring with n-pentane and drying, the solid, yellow residue was suspended in THF, filtered and examined by NMR spectroscopy. After repeating these three steps several times, 0.35 g (0.73 mmol/14%) of product was obtained in which the rac form, according to  $^1\text{H}$ -NMR, was enriched to more than 17:1.

Compound (9) exhibited a correct elemental analysis and the following NMR signals ( $\text{CDCl}_3$ , 100 MHz):  $\delta$  = 1.25 (s, 6H, Si-Me); 2.18 (s, 6H, 2-Me); 6.8 (s, 2H, 3-H-Ind); 6.92-7.75 (m, 8H, 4-7-H-Ind).

Supporting Catalyst Compound (9)

To an eight-liter vessel equipped with a cooling jacket and an efficient overhead stirrer was added methylalumoxane (30 wt% in toluene, 925 ml). With stirring, a suspension of compound 9 (10 g) in toluene (700 ml) was added under  $\text{N}_2$  through a double-ended needle. After stirring for 10 minutes, dehydrated silica (Davison 948, dried at 800°C, 200 g) was added to the solution over 20 minutes. The slurry was stirred for 10 minutes and then, while a vacuum was applied from the top of the vessel, a slight flow of  $\text{N}_2$  was added through the bottom. The mixture was heated to 70°C as the solvent was evaporated over a 5 hour period. The dry solid was cooled to ambient temperature overnight. Isopentane (6.4 liters) was added to slurry the solids and the mixture cooled to 0°C. Ethylene was added to the stirred mixture by a dip tube at a rate of 2.5-3.4 liters/minute until a total of 392 liters of ethylene had been added. Agitation was stopped and the solids allowed to settle. The liquid was decanted from the solids, which were washed twice, each with 3 liters of isopentane. The wet solids were transferred to a dry-



box under N<sub>2</sub> and filtered through a #14 mesh sieve. The sample was filtered, washed with pentane (4 liters) and dried in vacuo. Yield: 721 g.

Polymerization with Supported Compound 9

The polymerization procedure of compound 6 was followed using 199 mg of supported compound 9. Inspection of the interior of the reactor revealed it to be clean and free of fouling. The yield of isotactic polypropylene was 143 g.  $M_w = 172000$ , bulk density =  $0.36\text{g/cm}^3$ , average particle size = 802 microns, MP =  $147.9^\circ\text{C}$ .

10 C. Synthesis of rac-dimethylsilandiylobis(2-methyl-4,6-diisopropylindenyl)-zirconium dichloride

Synthesis of 2-Methyl-5, 7-diisopropyl-1-indanone (10) and 2-methyl-4,6-diisopropyl-1-indanone (11)

174 g (1300 mmol) of AlCl<sub>3</sub> were slowly added to a solution of 84.8 g (523 mmol) of 1,3-diisopropylbenzene and 120 g (523 mmol) of 2-bromoisobutyryl bromide (98% pure) in 600ml of analytical grade methylene chloride via a solids metering funnel at room temperature. The mixture was heated under reflux for a further 20 hours and then worked up analogously to Example A. The crude product was chromatographed on 3 kg of silica gel 60. The indanones 10 and 11 were able to be eluted separately with a mobile phase mixture of hexane/15% ethyl acetate. The compound 2-methyl-5-isopropyl-1-indanone followed as a by-product in a further zone with the same mobile phase. However, separation of the two isomers is not necessary for the further reactions. The overall yield was 93 g (78%). <sup>1</sup>H-NMR spectrum (360 MHz, CDCl<sub>3</sub>): isomer mixture (3:2) 7.49 (d), 7.36 (d), 7.13 (s), 7.10 (s), 4.15 (septet), 3.25-3.40 (m), 3.10 (septet), 2.90-3.00 (m), 2.60-2.73 (m), 1.22-1.30 (m). Mass spectrum: 230 M<sup>+</sup>, correct disintegration pattern.

Synthesis of 2-Methyl-4, 6-diisopropylindene (12) and 2-methyl-5,7-diisopropylindene (13)

19.3 g (511 mmol) of NaBH<sub>4</sub> were added to a solution of 78.5 g (341 mmol) of the isomer mixture of 10 and 11 in 700 ml of a solvent mixture of tetrahydrofuran/analytical grade methanol (2:1) at room temperature. After the mixture had been stirred at room temperature for 2 hours, 120-130 ml of half concentrated HCl were added, and the mixture was extracted with ether. The combined organic phases were dried with Na<sub>2</sub>SO<sub>4</sub>. The residue which remained

after the solvent had been stripped off was taken up in 500 ml of methylene chloride, and the mixture was heated under reflux with 6.5 g (34 mmol) of p-toluenesulfonic acid for 15 minutes. The residue which remained after the solvent had been stripped off was chromatographed on 1.5 kg of silica gel 60. 56 g of the isomer mixture of 12 and 13 were able to be isolated in the form of a colorless oil with a mobile phase mixture of hexane/diisopropyl ether 20:1. The overall yield was 86%.

<sup>1</sup>H-NMR spectrum (100 MHz, CDCl<sub>3</sub>): double bond isomers (1:1) 7.1 (m), 6.95 (m), 6.60 (m), 6.43, 3.25 (br), 2.75-3.20 (m), 2.12 (d), 1.28 (d), 1.25 (d).

Mass spectrum: 214 M<sup>+</sup>, correct disintegration pattern.

Synthesis of Dimethylbis (2-methyl-4, 6-diisopropylindenyl) silane (14)

9.2 ml (22.8 mmol) of a 2.5 M butyllithium solution in hexane were added to a solution of 4.9 g (22.8 mmol) of the isomer mixture of 12 and 13 in 25 ml of tetrahydrofuran at 0°C under Ar as an inert gas, and the mixture was heated under reflux for a further hour. The red solution was then added dropwise to a solution of 1.5 g (11.4 ml) of dimethyldichlorosilane in 10 ml of tetrahydrofuran at room temperature, and the mixture was heated under reflux for 8 hours. The batch was poured onto ice-water and extracted with ether. The ether phase was dried over magnesium sulfate and evaporated under reduced pressure. The yellowish oil which remained was then chromatographed on 500 g of silica gel 60. 1.4 g of the indene mixture of 12/13 were able to be eluted first with a mobile phase mixture of hexane/5% methylene chloride. The ligand system followed with hexane/8% methylene chloride. The viscous oil which remained after the mobile phase had been stripped off was able to be crystallized by stirring with methanol in an ice-bath. 3.1 g of a yellowish solid were obtained. The yield was 56%, or 84% with respect to the indene reacted.

<sup>1</sup>H-NMR spectrum (100 MHz, CDCl<sub>3</sub>): double bond isomers (3:1) 6.82-7.32 (m), 6.70 (m), 6.62 (m), 6.52 (m), 3.75 (s,br), 3.65 (s,br), 3.35 (s), 2.70-3.30 (m), 2.05-2.25 (m), 1.10-1.45 (m), 0.10-0.22 (m), -0.15 to -0.32 (m).

Mass spectrum: 484 M<sup>+</sup>, correct disintegration.

Synthesis of rac-Dimethylsilandiylbis-(2-methyl-4,6-diisopropylindenyl)-zirconium dichloride (15)

6.3 ml (16.2 mmol) of a 2.5 M butyllithium solution in hexane were added to a solution of 3.1 g (6.5 mmol) of the ligand system of 14 in 25 ml of diethyl ether at room temperature under Ar as an inert gas, and the mixture was stirred

overnight. After addition of 10 ml of hexane, the beige-colored suspension was filtered and the residue was washed with 20 ml of hexane. The dilithium salt was dried under an oil pump vacuum for a long time and then added to a suspension of 1.6 g (6.8 mmol) of  $ZrCl_4$  in 30 ml of methylene chloride at  $-78^\circ C$ . The mixture was warmed to room temperature in the course of 1 hour, and stirred at this temperature for a further 30 minutes. After the solvent had been stripped off, the orange-brown residue was extracted with 50 ml of hexane. After the solvent had been stripped off, 2.6 g (60%) of the complex 6 were obtained in the form of a yellow powder. The ratio of the racemate to the meso form was 3:1. 1.3 g (30%) of the complex 15 were obtained as the pure racemate (yellow crystalline powder) by recrystallization from hexane.

$^1H$ -NMR spectrum (100 MHz,  $CDCl_3$ ): 7.27 (2,s,aromatic-H), 7.05 (2,s,aromatic-H), 6.80 (2,s,b-Ind-H), 2.6-3.2 (4,m,i-Pr-CH), 2.22 (6,s,Ind-CH<sub>3</sub>), 1.15-1.40 (30,m,i-Pr-CH<sub>3</sub>,Si-CH<sub>3</sub>). Mass spectrum: 642 M<sup>+</sup> (with respect to  $^{90}Zr$ ), correct isotope pattern, correct disintegration.

#### Supporting Catalyst Compound 15

A solution of complex 15 (102 mg) dissolved in toluene (7 ml) was added dropwise to a solution of 30% methylalumoxane in toluene (10 ml). After the solution was stirred for 10 minutes at room temperature, Davison 948 silica (dried at  $800^\circ C$ , 2.0 g) was added over 5 minutes. Additional toluene (5 ml) was added to wash down the sides of the reaction flask. After 15 minutes, the solvent was evaporated in vacuo while heating to  $90^\circ C$ . The yield of catalyst was 4.44 g.

#### Polymerization with Supported Compound 15

44 mg of supported compound 15 slurried in hexane (2 ml) was flushed with propylene (250 ml) into a 2 liter autoclave, previously flushed with nitrogen, heated to  $65^\circ C$  and containing triethylaluminum (0.5 ml of a 1 M solution in hexane) and polypropylene (1000 ml). The reaction was run for 1 hour. The autoclave was cooled and vented and the polymer product dried in vacuo. The yield of propylene was 75 g. This material had a melt-flow rate of 0.6, a weight average molecular weight of 438,000, and a molecular weight distribution of 2.17.

#### D. Synthesis of rac-Dimethylsilandiyl-bis(2-methyl-4-phenyl-indenyl)zirconium dichloride Synthesis of (+)-2-(2-Phenyl-benzyl)propionic acid (16)

To 6.5 g (0.285 mol) sodium in 160 ml dry ethanol 48.6 g (0.279 mol) diethyl-methylmalonate was added at room temperature. 70.4 g (0.285 mol) 2-

phenyl-benzyl bromide was added to the stirred reaction mixture at such a rate, that the heat of reaction maintained the system at gentle reflux. Stirring at reflux was continued for 3 hours. 200 ml of water and 56 g (1 mol) potassium hydroxide were added and the reaction mixture was heated to reflux for 4 hours. The solvents were distilled off, water was added until the residue was completely dissolved and concentrated HCL was added to adjust the pH to 1. The precipitate was collected on a filter, dried and heated in a flask to 130°C. The acid then employed as is. Yield (58 g, 85%). <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 100 MHz): 11.7 (s, 1H, COOH), 7.1 - 7.5 (m, 9H, arom. H), 2.3 - 3.2 (m, 3H, CH and CH<sub>2</sub>), 0.9 (d, 3H, CH<sub>3</sub>).

#### Synthesis of (+)-2-Methyl-4-phenyl-1-indanone (17)

A solution of 58 g (0,242 mol) of 16 in 60 ml (0,83 mol) thionyl chloride was stirred 18 hr at room temperature. The excess of thionyl chloride was distilled off, 100 mls of toluene were added three times and removed in vacuo.

The solution of the acid chloride in 150 ml dry toluene was added at 10°C to a stirred suspension of 48 g (0.363 mol) aluminum trichloride in 400 ml dry toluene over a period of 1 hour. After heating to 80°C for 3 hours the mixture was poured over 500 g ice, acidified with concentrated HCL to pH = 1 and extracted with 200 ml diethyl ether three times. The combined organic layers were washed with a saturated aqueous NaHCO<sub>3</sub>-solution and brine. The solvents were removed under vacuo. The indanone 17 was used without further purification. Yield (50 g, 93%)

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 100 MHz): 7.2-7.8 (m, 8H, arom. H), 3.3 (dd, 1H, H-C(3), 2.5-2.9 (m, 2H, H'-C(3) and H-C(2)), 1.3 (dd, 3H, CH<sub>3</sub>).

#### Synthesis of 2-Methyl-7-phenyl-indene (18)

50 g (0.226 mol) of 17 in 450 ml THF/methanol (2:1) was treated at 0°C with 12.8 g (0.34 mol) sodium borohydride in portions. The reaction mixture was stirred at room temperature for 16 hours. The mixture was poured on 500 g ice, acidified with concentrated HCL to pH = 1 and extracted three times with 200 ml diethyl ether. The combined organic layers were washed with brine and dried over MgSO<sub>4</sub>. After removal of the solvent the residue was heated to reflux in 1000 ml toluene containing 2 g p-toluenesulfonic acid monohydrate for 2 hours. To remove the acidic catalyst the mixture was washed with 200 ml of a saturated aqueous NaHCO<sub>3</sub>-solution. After removal of the solvent and chromatography over silica

gel (hexane/CH<sub>2</sub>Cl<sub>2</sub> 9:1) the indene 18 was obtained as a colorless oil. Yield (42 g, 90%).

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 100 MHz): 7.0 -7.6 (m, 8H, arom. H), 6.5 (m, 1H, H-C(3)), 3.4 (s, 2H, CH<sub>2</sub>), 2.1 (sa, 3H, CH<sub>3</sub>).

5 Synthesis of Dimethyl-bis(2-methyl-phenyl-indenyl)silane (19)

To a solution of 15 g (73 mmol) of 18 in 200 ml dry toluene and 10 ml dry THF under an argon atmosphere 29 ml (73 mmol) of a 2.5 M solution of butyllithium in hexane was added at room temperature. After stirring at 80°C for 1 hour the reaction mixture was cooled to 0°C and 4.7 g (36.5 mmol) dimethylsilane  
10 dichloride was added. The mixture was heated to 80°C for 1 hour and after the reaction was complete 100 ml of water was added. The organic layer was separated, the solvent was removed under vacuo and the residue purified by chromatography over silica gel (hexane/CH<sub>2</sub>Cl<sub>2</sub> 9:1). Yield (12 g, 70%).

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 100 MHz). 7.1 -7.7 (m, 16H, arom. H), 6.8 (m, 2H, H-C(3)),  
15 3.8 (s, 2H, H-C(1)), 2.2 (s, 6H, CH<sub>3</sub>), -0.2 (m, 6H, CH<sub>3</sub>Si).

Synthesis of rac-Dimethylsilandiyl-bis(2-methyl-4-phenyl-indenyl)zirconium dichloride (20)

To a solution of 6g (13 mmol) of 19 in 100 ml dry toluene under an argon atmosphere 10.6 ml (26 mmol) of a 2.5 M solution of butyllithium in hexane was  
20 added at room temperature. After stirring at reflux for 3 hour the reaction mixture was cooled to -25°C and 3.2 g (13.6 mmol) zirconium tetrachloride was added. The reaction mixture was allowed to warm to room temperature over a period of 2 hours. The suspension was filtered over a G3-schlenck frit and the residue was washed with 50 ml dry toluene. The solvent of the combined extracts was  
25 removed under vacuo. 9.0 g of the complex were obtained as a 1:1 mixture of the racemic and the meso-form. 2.74 g (33%) of the pure racemate were obtained after fractional crystallization of the 1:1-mixture from dry CH<sub>2</sub>Cl<sub>2</sub>. <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 100 MHz): 7.0 - 7.7 (m, 16 H, arom. H), 6.9 (s, 2H, H-C(3)), 2.2 (s, 6H, CH<sub>3</sub>), 1.3(m, 6H, CH<sub>3</sub>Si). MS (EI) 626 (<sup>90</sup>Zr<sup>35</sup>Cl<sub>2</sub>).

30 Supporting Catalyst Compound 20

To a filtered solution of compound 20 (0.05 g) in toluene (75 ml) the MAO slurry (30% MAO in toluene, 5.3 ml) was added with stirring. After five minutes the solution was red-orange and slightly hazy. To this was added dehydrated silica (6.0 g, Davison 948 regular, 800°C dehydration) and stirred fifteen minutes. The  
35 resultant slurry was evaporated on a rotoevaporator at 44°C over twenty five

minutes at which point the slurry had reached to a 'mud' stage. After an additional thirty five minutes drying at 46°C the solids were recovered as a light orange solid (7.2g)

Polymerization with Compound 20

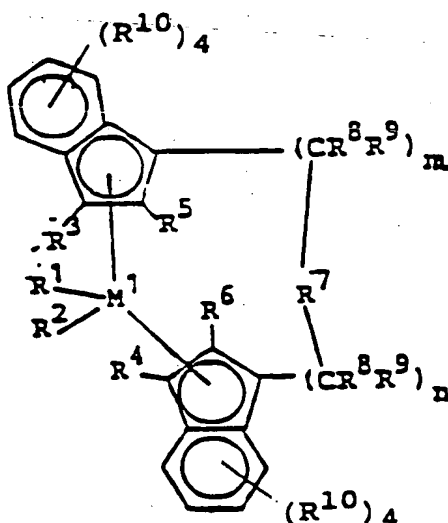
5 In a clean, dry two liter autoclave which had been flushed with propylene vapor, TEAL (0.8 ml, 1.5 M in heptane) was added and then the reactor closed and filled with 750 ml liquid propylene. With the reactor temperature at 30°C the supported catalyst 20 (as a wt% oil slurry) was washed in via an addition tube with 250 ml propylene. The reactor was rapidly heated to 65°C. After thirty minutes  
10 the reactor was cooled and the excess propylene vented. Inspection of the interior of the reactor revealed it to be clean and free of fouling. Isotactic polypropylene was removed and dried. Catalyst activity = 0.37 kg/g/hr, DSC Mp = 150°C, weight average MW = 748000 g/m, MWD = 2.2.

15 This completes the description of the preferred and alternate embodiments of the invention. Those skilled in the art may recognize other equivalents to the specific embodiment described herein which equivalents are intended to be encompassed by the claims attached hereto.

## Claims:

1. A supported catalyst system comprising:

- (a) reaction product of one or more metallocene(s) and alumoxane, said reaction product obtained by contacting said metallocene component(s) and alumoxane in a suitable solvent, said metallocene represented by the general formula:



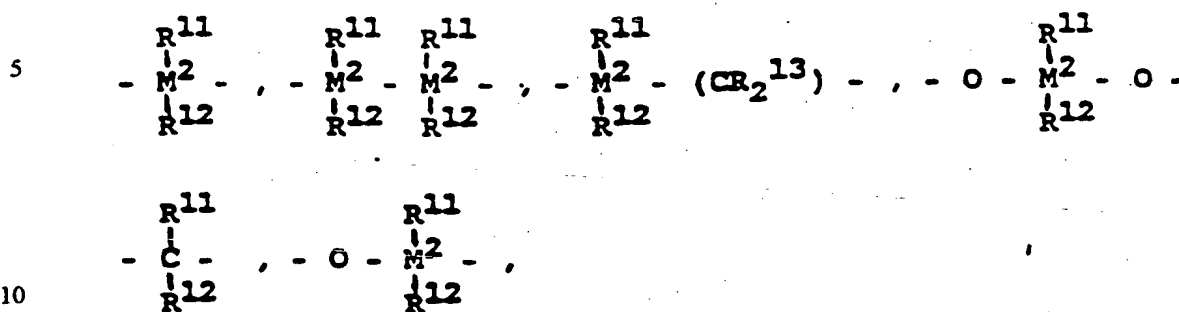
wherein:  $M^1$  is a metal of group 4, 5, or 6 of the Periodic Table,

$R^1$  and  $R^2$  are identical or different, are one of a hydrogen atom, a  $C_1$ - $C_{10}$  alkyl group, a  $C_1$ - $C_{10}$  alkoxy group, a  $C_6$ - $C_{10}$  aryl group, a  $C_6$ - $C_{10}$  aryloxy group, a  $C_2$ - $C_{10}$  alkenyl group, a  $C_7$ - $C_{40}$  arylalkyl group, a  $C_7$ - $C_{40}$  alkylaryl group, a  $C_8$ - $C_{40}$  arylalkenyl group, or a halogen atom;

$R^3$  and  $R^4$  are hydrogen atoms;

$R^5$  and  $R^6$  are identical or different, preferably identical, are one of a halogen atom, a  $C_1$ - $C_{10}$  alkyl group, which may be halogenated, a  $C_6$ - $C_{10}$  aryl group, which may be halogenated, a  $C_2$ - $C_{10}$  alkenyl group, a  $C_7$ - $C_{40}$  arylalkyl group, a  $C_7$ - $C_{40}$  alkylaryl group, a  $C_8$ - $C_{40}$  arylalkenyl group, a  $-NR_2^{15}$ ,  $-SR^{15}$ ,  $-OR^{15}$ ,  $-OSiR_3^{15}$  or  $-PR_2^{15}$ -radical, wherein  $R^{15}$  is one of a halogen atom, a  $C_1$ - $C_{10}$  alkyl group, or a  $C_6$ - $C_{10}$  aryl group;

- 30 -

R<sup>7</sup> is

=BR<sup>11</sup>, =AIR<sup>11</sup>, -Ge-, -Sn-, -O-, -S-, =SO, =SO<sub>2</sub>, =NR<sup>11</sup>, =CO, PR<sup>11</sup>, or =P(O)R<sup>11</sup>;

15 wherein:

R<sup>11</sup>, R<sup>12</sup> and R<sup>13</sup> are identical or different and are a hydrogen atom, a halogen atom, a C<sub>1</sub>-C<sub>20</sub> alkyl group, a C<sub>1</sub>-C<sub>20</sub> fluoroalkyl group, a C<sub>6</sub>-C<sub>30</sub> aryl group, a C<sub>6</sub>-C<sub>30</sub> fluoroaryl group, a C<sub>1</sub>-C<sub>20</sub> alkoxy group, a C<sub>2</sub>-C<sub>20</sub> alkenyl group, a C<sub>7</sub>-C<sub>40</sub> arylalkyl group, a C<sub>8</sub>-C<sub>40</sub> arylalkenyl group, C<sub>7</sub>-C<sub>40</sub> alkylaryl group, or R<sup>11</sup> and R<sup>12</sup>, or R<sup>11</sup> and R<sup>13</sup>, together with the atoms binding them, can form ring systems;

M<sup>2</sup> is silicon, germanium or tin;

R<sup>8</sup> and R<sup>9</sup> are identical or different and have the meanings as stated for R<sup>11</sup>;

25 m and n are identical or different and are zero, 1 or 2, m plus n being zero, 1 or 2; and

the radicals R<sup>10</sup> are identical or different and have the meanings stated for R<sup>11</sup>, R<sup>12</sup> and R<sup>13</sup> wherein further two adjacent R<sup>10</sup> radicals can be joined together to form a ring system;

30 (b) a porous support; and,

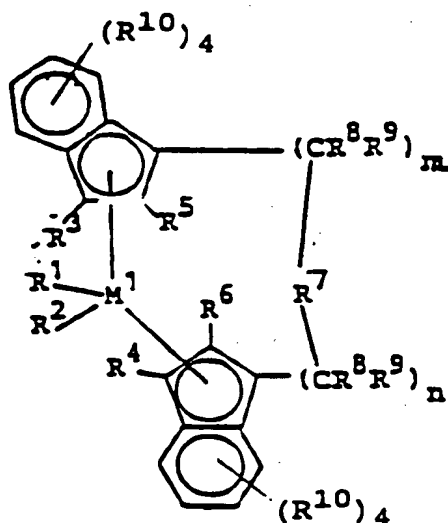
(c) optionally prepolymerized olefinic monomer(s).

2. A process for the preparation of a homo- or copolymer of propylene which has a molecular weight of about 100000 grams/mole or greater, a melting point of about 140°C or greater, a bulk density of about 0.30 g/cm<sup>3</sup> or

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greater by polymerization of propylene or copolymerization of mixtures of propylene and one or more other olefins, at a polymerization reaction temperature of about 45°C or greater, at a polymerization pressure of from 0.5 - 100 bar, in gas phase, bulk, or slurry polymerization in the presence of  
 5 (a) reaction product of one or more metallocene(s) and alumoxane, said reaction product obtained by contacting said metallocene component(s) and alumoxane in a suitable solvent, said metallocene represented by the general formula:



wherein:  $M^1$  is a metal of group 4, 5, or 6 of the Periodic Table,

$R^1$  and  $R^2$  are identical or different, are one of a hydrogen atom, a  $C_1$ - $C_{10}$  alkyl group, a  $C_1$ - $C_{10}$  alkoxy group, a  $C_6$ - $C_{10}$  aryl group, a  $C_6$ - $C_{10}$  aryloxy group, a  $C_2$ - $C_{10}$  alkenyl group, a  $C_7$ - $C_{40}$  arylalkyl group, a  $C_7$ - $C_{40}$  alkylaryl group, a  $C_8$ - $C_{40}$  arylalkenyl group, or a halogen atom;

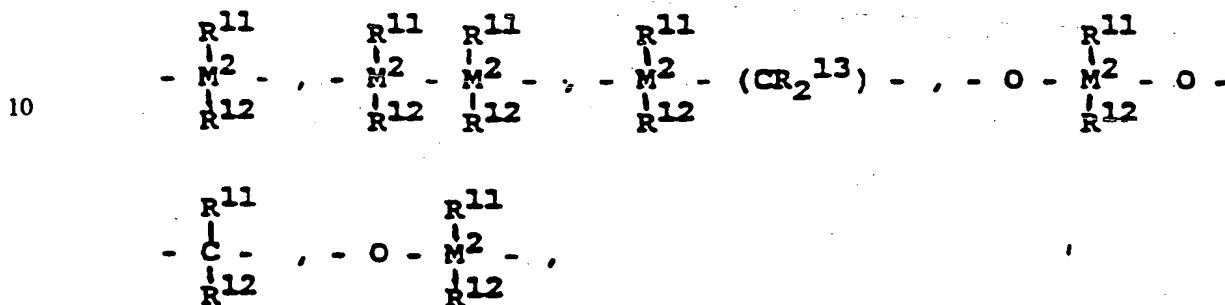
$R^3$  and  $R^4$  are hydrogen atoms;

$R^5$  and  $R^6$  are identical or different, preferably identical, are one of a halogen atom, a  $C_1$ - $C_{10}$  alkyl group, which may be halogenated, a  $C_6$ - $C_{10}$  aryl group, which may be halogenated, a  $C_2$ - $C_{10}$  alkenyl group, a  $C_7$ - $C_{40}$  arylalkyl

group, a C<sub>7</sub>-C<sub>40</sub> alkylaryl group, a C<sub>8</sub>-C<sub>40</sub> arylalkenyl group, a -NR<sup>15</sup>, -SR<sup>15</sup>, -OR<sup>15</sup>, -OSiR<sub>3</sub><sup>15</sup> or -PR<sub>2</sub><sup>15</sup> radical, wherein R<sup>15</sup> is one of a halogen atom, a C<sub>1</sub>-C<sub>10</sub> alkyl group, or a C<sub>6</sub>-C<sub>10</sub> aryl group;

R<sup>7</sup> is

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=BR<sup>11</sup>, =AlR<sup>11</sup>, -Ge-, -Sn-, -O-, -S-, =SO, =SO<sub>2</sub>, =NR<sup>11</sup>, =CO, PR<sup>11</sup>, or

20 =P(O)R<sup>11</sup>;

wherein:

R<sup>11</sup>, R<sup>12</sup> and R<sup>13</sup> are identical or different and are a hydrogen atom, a halogen atom, a C<sub>1</sub>-C<sub>20</sub> alkyl group, a C<sub>1</sub>-C<sub>20</sub> fluoroalkyl group, a C<sub>6</sub>-C<sub>30</sub> aryl group, a C<sub>6</sub>-C<sub>30</sub> fluoroaryl group, a C<sub>1</sub>-C<sub>20</sub> alkoxy group, a C<sub>2</sub>-C<sub>20</sub> alkenyl group, a C<sub>7</sub>-C<sub>40</sub> arylalkyl group, a C<sub>8</sub>-C<sub>40</sub> arylalkenyl group, C<sub>7</sub>-C<sub>40</sub> alkylaryl group, or R<sup>11</sup> and R<sup>12</sup>, or R<sup>11</sup> and R<sup>13</sup>, together with the atoms binding them, can form ring systems;

M<sup>2</sup> is silicon, germanium or tin;

R<sup>8</sup> and R<sup>9</sup> are identical or different and have the meanings as stated for

30 R<sup>11</sup>;

m and n are identical or different and are zero, 1 or 2, m plus n being zero,

1 or 2; and

the radicals R<sup>10</sup> are identical or different and have the meanings stated for R<sup>11</sup>, R<sup>12</sup> and R<sup>13</sup> wherein further two adjacent R<sup>10</sup> radicals can be joined together

35 to form a ring system;

- 33 -

- (b) a porous support; and
- (c) optionally prepolymerized olefinic monomer(s).

3. The catalyst system of claim 1 or 2 prepolymerized with olefinic monomer.
- 5 4. The catalyst system of any preceeding claims wherein  $m=n=0$  and  $M^2=\text{silicon}$ .
- 10 5. The catalyst system of any preceeding claims wherein  $R^5 = R^6 = C_{1-10}$  alkyl.
6. The catalyst system of any preceeding claims wherein three of the  $R^{10}$  radicals = hydrogen and one is a  $C_{6-30}$  aryl group.
- 15 7. The catalyst system of any preceeding claims wherein two adjacent  $R^{10}$  radicals form a fused 4,5-benzo ring and the other two  $R^{10}$  radicals are hydrogen.
- 20 8. The catalyst system of any preceeding claims wherein at least one of the  $R^{10}$  radicals is  $C_{1-10}$  alkyl.
- 25 9. The catalyst system of any preceeding claims wherein the metallocene component is selected from the group consisting essentially of:  
rac-dimethylsilandiylbis(2-methyl-4,5-benzoindenyl) - zirconium dichloride;  
rac-dimethylsilandiylbis(2-methylindenyl)-zirconium dichloride;  
rac-dimethylsilandiylbis(2-methyl-4,6-diisopropylindenyl)-zirconium dichloride; and  
rac-dimethylsilandiylbis(2-methyl-4-phenylindenyl)zirconium dichloride.
- 30 10. The catalyst system of any preceeding claims wherein the olefinic monomer(s) comprises ethylene.
11. The catalyst system of claims 1-9 wherein the olefinic monomer(s) comprises propylene.

12. The catalyst system of claims 1-9 wherein the olefinic monomer(s) comprises at least two different monomers having from 2-8 carbon atoms.
13. The catalyst system of any preceeding claims wherein the olefinic monomer(s) comprises from about 10% to about 1000% of the weight of the non-prepolymerized supported catalyst weight.
14. The catalyst system of claim 13 wherein the olefinic monomer(s) comprises about 100% to about 300% of the weight of the non-prepolymerized supported catalyst weight.
15. The catalyst system of any preceeding claims wherein the molar ratio of aluminum, of the alumoxane, to metal, of the metallocene, is about 12:1 to about 1000:1.
16. The catalyst of claim 15 wherein the ratio is about 50:1 to about 500:1.
17. The process of claim 2 wherein propylene is polymerized to isotactic polypropylene.
18. The process of claims 2, or 17 wherein mixtures of propylene and other polymerizable olefins are polymerized to copolymers of polypropylene.
19. Isotactic polymers of propylene made in accordance with the process of claims 2, 17 or 18 comprising a weight average molecular weight of about 100,000 grams/mole or greater, a melting point of about 140°C or greater, and a bulk density of about 0.30 grams/cm<sup>3</sup> or greater.
20. Isotactic polypropylene produced in the presence of the catalyst system of any of claims 1-16.
21. A method to produce a supported catalyst system comprising the steps of:
- a) forming in a suitable solvent, a metallocene/alumoxane mixture wherein the metallocene component is in accordance with claim 1 or 3-16,
  - b) contacting the mixture in a suitable solvent with a porous carrier;

- c) removing essentially all the solvent from the resulting slurry;
- d) obtaining a supported catalyst system; and
- e) optionally prepolymerizing the supported catalyst system with one or more olefinic monomer(s).

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22. The process of claim 2 wherein the catalyst system is prepolymerized to produce polypropylene polymer or copolymers thereof having a molecular weight about 100000 or greater, a melting point of about 140°C or greater, and a bulk density of about 0.30 g/cm<sup>3</sup> or greater.

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23. The use of the polymers produced from the process of claims 2, or 17-20 in shaped articles, films or fibers.

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FIG. 1

Metallocene		Silico		Catalyst Preparation				Analytical Analysis		Calculated Quantities		Polymerization Performance RxnT-65°C 0.5 mm TEAL Scavenger											
Name	Amt. (g)	Temp. (c)	Amt. (g)	Conc. Amt. (wt%) (mls)	Prepoly Monomer	Amt. (SCF)	Temp. (F)	Add'n Rate (SCFM)	Yield (g)	Si (wt%)	Al (wt%)	Zr (wt%)	Al/Zr (molar)	MC (wt%)	MWD	Pol Yield (g)	Cat Amt (mg)	Cot Act (gpol/ g cat/hr)	MP (°C)	MW (g/m) (K)	BD g/cc	avg. PS (μ)	
A	5	800	200	30	925	C <sub>2</sub> =	9.7	30-40	0.03-0.06	326.2	12.69	9.86	0.079	422	0.41	1.90	305	199	1533	146	273	43	750
B	10	800	200	30	925	C <sub>2</sub> =	13.9	26-60	0.08-0.12	721.6	10.74	7.88	0.19	140	1.00	2.02	143	199	720	148	172	36	802
C	-	800	2.0	-	102	-	-	-	-	4.44	-	-	-	-	-	2.17	75	44	-	-	438	-	-
D	-	800	6.0	-	50	-	-	-	-	7.2	-	-	-	-	-	2.2	-	-	-	150	748	-	-

## INTERNATIONAL SEARCH REPORT

 Inte. onal Application No  
 PCT/U /05636

 A. CLASSIFICATION OF SUBJECT MATTER  
 IPC 5 C08F4/602 C08F10/00

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 5 C08F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
P, X	EP, A, 0 563 917 (HOECHST AKTIENGESELLSCHAFT) 6 October 1993 see example 5 ---	1, 4, 5, 9, 21
A	EP, A, 0 485 822 (HOECHST AKTIENGESELLSCHAFT) 20 May 1992 see page 2, line 1 - line 2 see examples 1-14 see claims 1-5 ---	1
P, X	EP, A, 0 576 970 (HOECHST AKTIENGESELLSCHAFT) 5 January 1994 see page 15; example A see example 26 ---	1, 2, 4, 6, 9
P, X	EP, A, 0 566 349 (MITSUI PETROCHEMICAL INDUSTRIES) 20 October 1993 see example 7 ---	1, 3-5, 9, 11, 13, 14
	--- -/-	

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

## \* Special categories of cited documents :

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- \* "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
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- \* "P" document published prior to the international filing date but later than the priority date claimed

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## INTERNATIONAL SEARCH REPORT

International Application No

PCT/US92/05636

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP,A,0 518 092 (BASF AKTIENGESELLSCHAFT) 16 December 1992 see page 3, line 30 - line 42 see claims 1,8,10 -----	1,4-9



# INTERNATIONAL SEARCH REPORT

information on patent family members

Int. application No

PCT/US94/05636

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A-0563917	06-10-93	AU-B- 3559893 JP-A- 6025348	07-10-93 01-02-94
EP-A-0485822	20-05-92	AU-B- 640286 AU-A- 8775991 CA-A- 2055216 JP-A- 4268307 US-A- 5243001	19-08-93 14-05-92 13-05-92 24-09-92 07-09-93
EP-A-0576970	05-01-94	AU-B- 4149193 CA-A- 2099214 JP-A- 6100579	06-01-94 28-12-93 12-04-94
EP-A-0566349	20-10-93	NONE	
EP-A-0518092	16-12-92	DE-A- 4119343 JP-A- 5140226	17-12-92 08-06-93